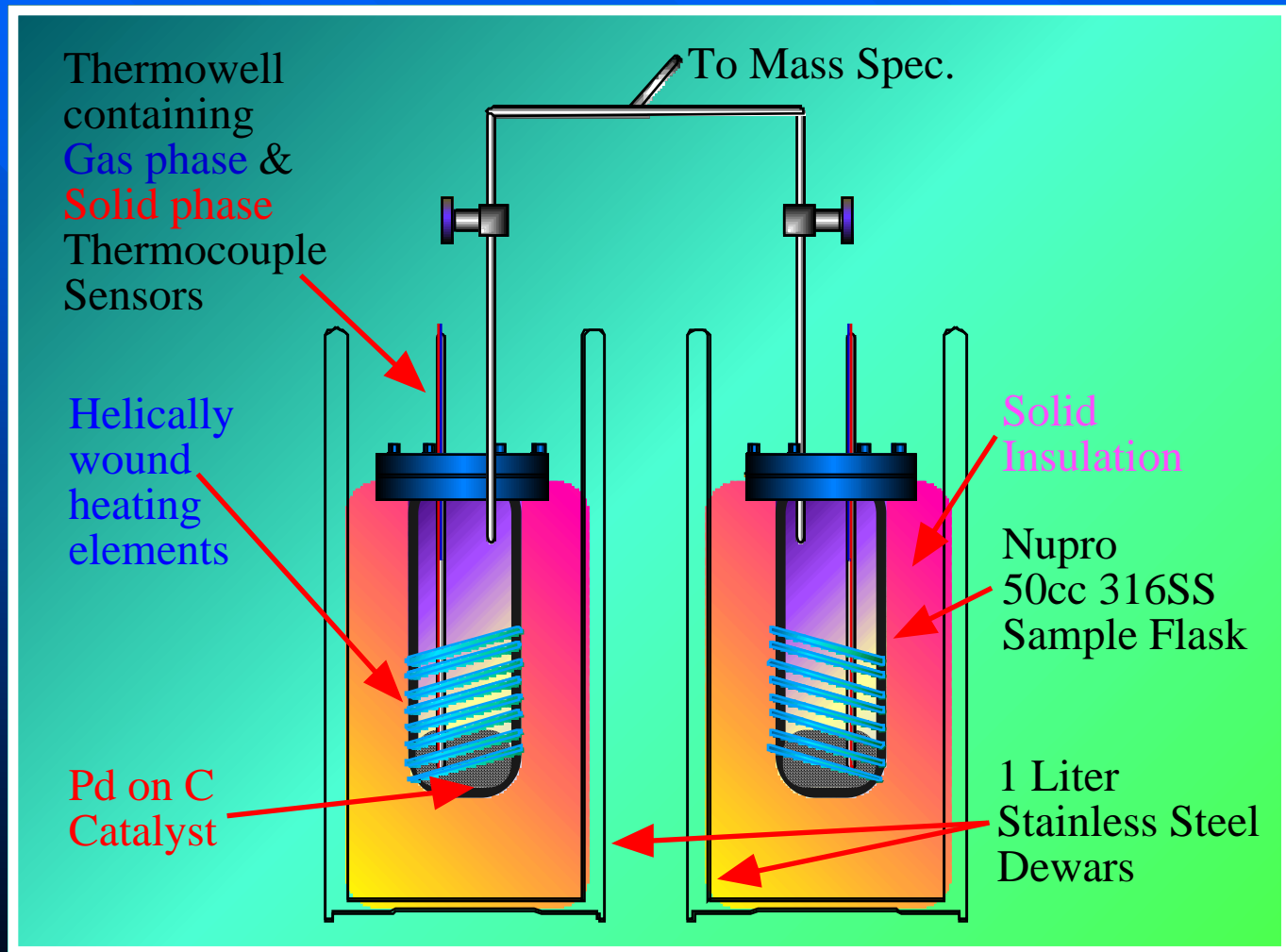


Review of experimental measurements involving dd reactions

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Case cell Studies: D₂ Gas with Pd/C Catalyst



Case Conclusions

- Near quantitative correlation between Heat and ^4He production according to:

Predicted: $d + d \rightarrow ^4\text{He} + \sim 24\text{MeV}_{(lattice)}$

Measured: $Q = 31 \pm 13 \text{ MeV/atom}$

- Discrepancy may be due to solid phase retention of ^4He
- Substantial initiation time $\gg D$ diffusion.
- $\text{Max } [^4\text{He}]_{\text{Sample}} / [^4\text{He}]_{\text{Air}} > 2$

Production of Tritium in a Sealed Pd cavity

AZ1 0.3M LiOD, AZ2 0.3M LiOH

Cathodic Current 5 - 7.5A

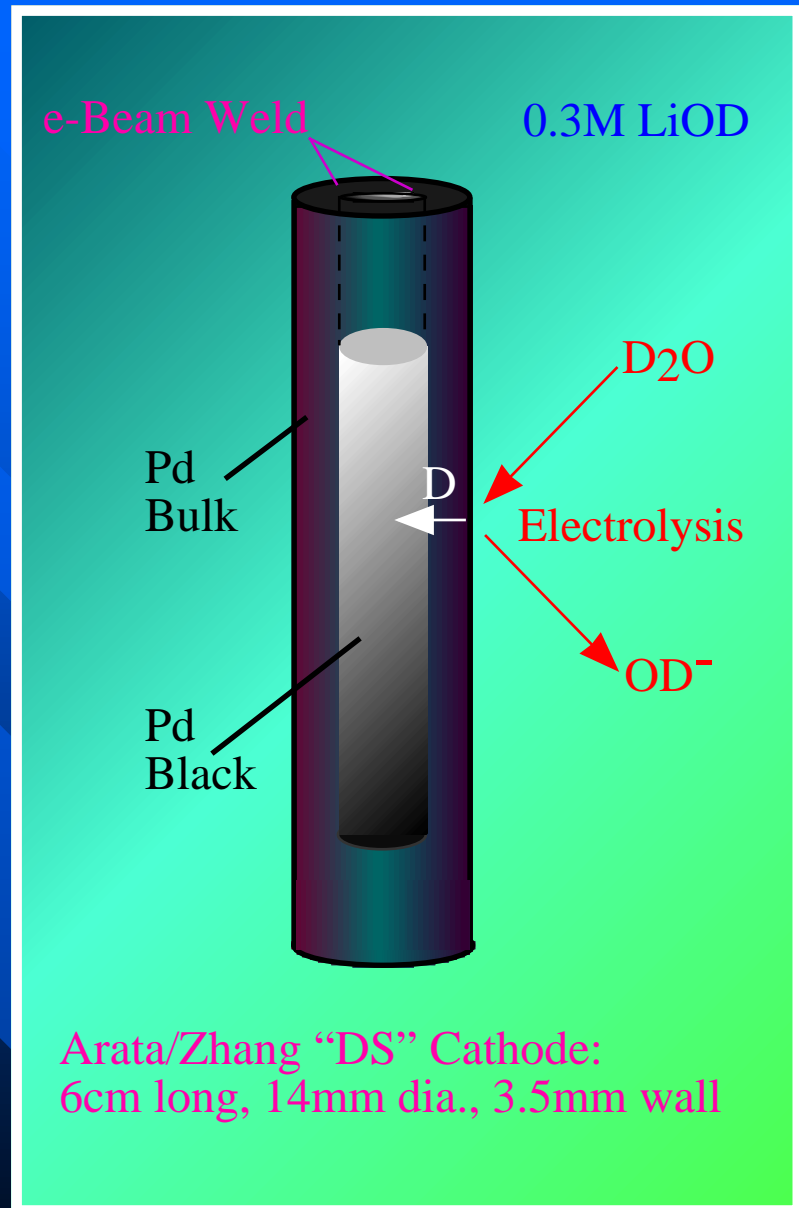
Current Density 170-255mA cm⁻²

P_{in} 50-317 W, Duration 120 Days

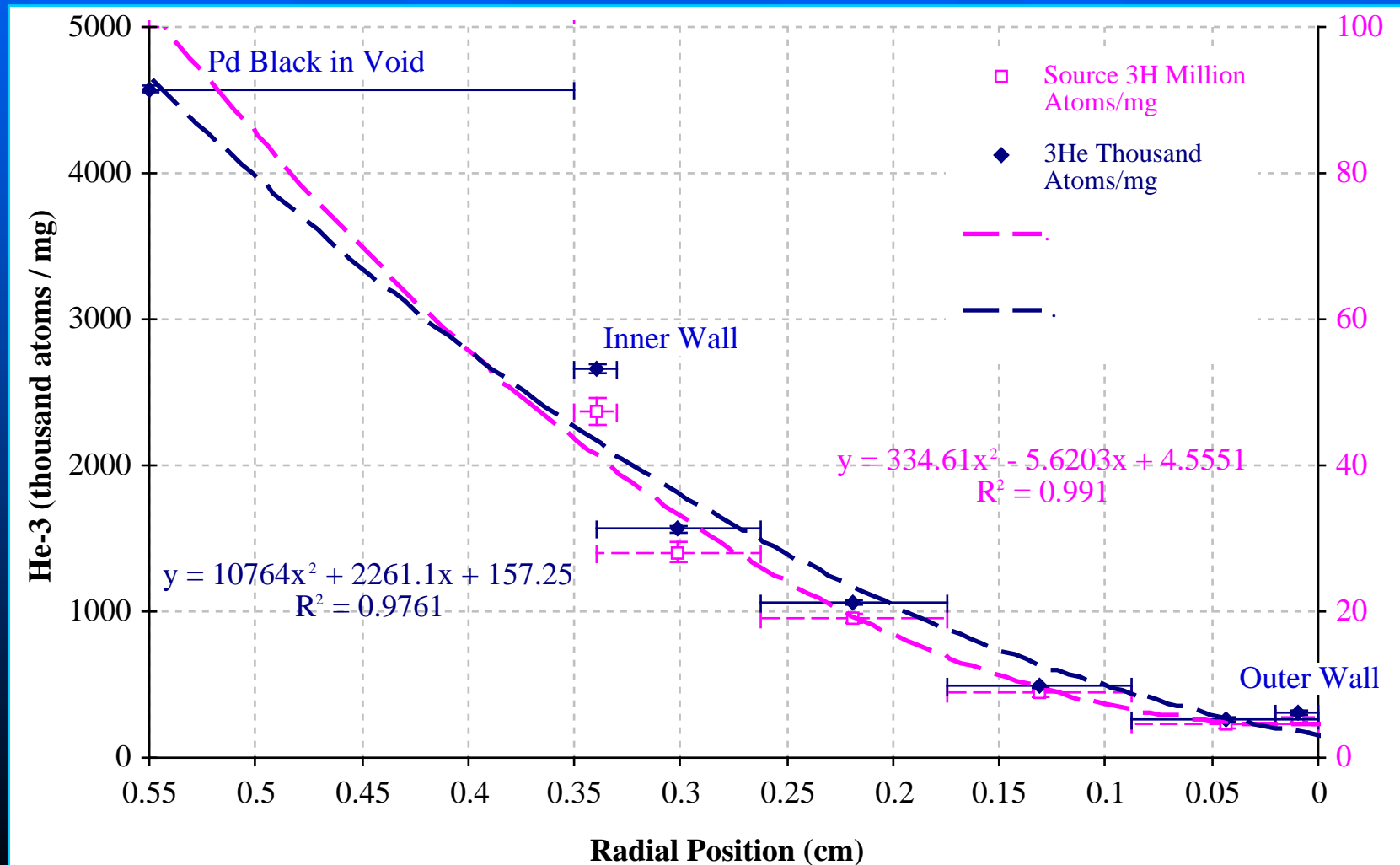
P_{xs,Max} = 10 ±1.5%, P_{xs} 0 ±1.5%,

Deloaded:

open circuit and at 2V Anodic
for a further 100 Days.



AZ1: Radial Distribution of ^3He and ^3H



Tritium Conclusions

- Production of Tritium was between 2×10^{15} and 5×10^{15} atoms.

Modeled as a single event, this occurred during cathodic electrolysis.

There is definite evidence of excess ^3He from Tritium decay of all samples of Pd & Pd-black from the D_2O experiment.

Samples of Pd taken from a similar and contemporaneous H_2O electrode show low ^3He levels consistent with blank Pd.

Measurements of the ^3He gradient through the 3.5mm wall of the D_2O electrode show that the ^3He is the decay product of Tritium which diffused from a source inside the electrode.

No evidence for ^4He quantitatively consistent with excess heat.