Test of Energy Catalyzer

Bologna April 19, 2011

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Measured voltage: 236 V (AC)

Weight hydrogen bottle (attached, opened, closed, and detached):

- before: 13653.1 grams - after: 13652.6 grams Total loaded: 0.5 grams

Pressure H2

Bottle: 85 bar Reduced: 12 bar

Temperature

- T2 temperature water inlet
- T3 temperature outlet

17.03	Water pump started.		
17.07	Water comes out of outlet hose.		
17.10	Water inlet from reservoir 2 – initial weight 10280.7 grams		
17.12	Start control unit. Measured AC current: 153 mA, which means 36 watts		
17.15	Start heating Measured AC current: 1.5 A, which means 354 watts T2: 20.9°C T3: 21.3°C		
17.21	Sound of heating water. T3: 67°C		
17.25	T=99.9 °C. Outlet hose hot.		
17.31	Change reservoir to 1 Reservoir 1: initial weight 10185.9 grams Reservoir 2: 8826.6 grams (consumed 1454.1 grams in 21 minutes, which means 4.15 kg/h). T2: 22.8°C T3: 100.0°C		
18.25	Added water B before: B after: Tot added:	from reservoir 2 into 1 8826.4 grams 6001.6 grams 2824.8 grams	T2: 22.6°C T2: 22.1°C
19.07	Added water B before: B after: Tot added:	from reservoir 2 into 1 6001.0 grams 2195.2 grams 3805.8 grams	T2: 22.0°C T2: 21.6°C
19.41	Stop. Reserve	oir 1: 7891.7 grams	T2: 21.6°C

Water flow

Reservoir 1

Weight at start 17.31	+10185.9 grams
Added 18.25	+ 2824.8 grams
Added 19.07	+ 3805.8 grams
Remaining 19.41	– 7891.7 grams
Tot consumed in 2:10 h	8924.8 grams, which means 4.12 kg/h

Energy calculation:

Conservative value of inlet water temperature, T2: 22.5°C Boiling temperature: 99,5°C ΔT = 77K

Heat capacity of water is 4.18 kJ/(kg x K)

Energy required for heating water, $W_{heat} = 321.86 \text{ kJ/kg} = 89.41 \text{ Wh/kg}$

Enthalpy of vaporization is 2260 kJ/kg. Energy required for vaporization, $W_{vap} = 627.8$ Wh/kg.

Total energy required, $W_{tot} = W_{heat} + W_{vap} = 717.2 \text{ Wh/kg}$

Total mass of water, $m_{tot} = 8924.8$ grams

Total energy produced $W_{tot} \ge m_{tot} = 717.2 \ge 8.9248 = 6401$ Wh

Max electric heating energy (including 36 watts power for control unit): $354 \times 2.167 = 767 \text{ Wh}$

Net energy: 5634 Wh \approx 5.6 kWh (\approx 20 MJ)

Average net heating power in 2:10 (2.167) h: 5.634 / 2.167 = 2.6 kW

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Temperature during start-up:



Instruments

* Peristaltic pump NSF Model # CEP183-362N3 Serial # 060550065 Max output 12.0 liters/h Max press 1.50 bar

* Temperature logger Testo 177-T3 0554 1765 Usb Interface

* Temperature probe Testo Calibration certificate No 838/2010 PD 29/12 2010 (measured T3) Calibration certificate No 839/2010 PD 29/12 2010 (measured T2)

* Radiation meter Automess S/N 95171

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* Scale Model: TKW 15 S S/N: 2917029003 Max 15000 g d= 0.1 g Certified according to ISO 9001:2000

* Amperemeter Digimaster DM2011090647637(Checked against a multimeter afterwards by Mats Lewan).

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