From Cold Fusion to Condensed Matter Nuclear Science

M. C. H. McKubre*, F. L. Tanzella* and V. Violante**

*Energy Research Laboratory SRI International, Menlo Park, CA, 94025, USA.

**ENEA, Frascati, Italy.

Michael.McKubre@SRI.com

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A brief history of "Cold Fusion"

- The name "cold fusion" appears to have originated with Jones and Rafelski.
- These authors invoked the phrase to describe muon catalyzed pairwise d-d fusion and later argued in 1989 that this effect was responsible for low level neutron generation in condensed matter reactions.
- The evidence for heat effects in palladium deuteride was clearly demonstrated by Fleischmann and Pons in 1989 and ultimately determined to be sound.
- Direct evidence of commensurate fusion product creation was slow in coming with Miles *et al* first claim of near quantitative ⁴He production in 1992.
- To this date no evidence has been accumulated of reactant consumption in d -d heat production.
- The measured product distributions, ${}^{4}\text{He} >> {}^{3}\text{H} >> n_{\circ}$, cannot be associated plausibly with two body fusion effects
- The effect takes advantage of <u>condensed states</u> of matter - possibly crystalline or surface states

Condensed Matter Nuclear Science

- The phrase "Condensed Matter Nuclear Science" or CMNS was crafted in May 2002 at the IAC meeting of ICCF9 chaired by Prof. Li:
 - to extend the reference topic area,
 - to incorporate explicitly the connection of the phenomenon with the condensed state,
 - and accommodate increasing evidence of nuclear products not consonant with orthodox fusion or fission reactions.
- This broadening in emphasis was both rational and necessary to accommodate new information.
- This change has defocused attention from the original claim of <u>PdD</u> <u>nuclear -level heat energy</u> (and possibly helium).
- This has led to two unforeseen consequences that are largely negative, at least in the short term:
 - (i) the parameter space of excess heat production has been insufficiently well studied and understood to institute a fully replicable experiment;
 - (ii) the practical utility of metal deuteride heat production is not yet well defined in it's limits or even application.

Energetics and SuperWaves[®]

- A program instituted by Energetics is seeking to help redress these two deficiencies in CMNS studies:
 - by controlling the palladium metallurgy,
 - by controlling surface morphology,
 - and particularly the loading and excitation waveform(s) of electrolytic cathodes based on the theoretical concepts of Dardik.
- A program recently completed at SRI was successful in replicating experiments performed initially by Energetics scientists in Omer, Israel.
- A second, independent replication attempt was mounted and successfully completed at ENEA, Frascati.
- Results of the work at SRI and ENEA will be discussed in the context of the replicability and practically of CMNS heat effects.

SuperWaves[®]



Energetics Isoperibolic Calorimeter and results



Energy Input 40 kJ, Output 1.14 MJ > 28 x E_{Out}/E_{In} , > 70 x P_{Out}/P_{In} .

Calibration



Summary of results

Cell -	Cathode	Min.	Max.	Excess	Power	Energy
Calorimeter		R/R°	D/Pd	$\%$ of $P_{\mbox{\tiny In}}$	(mW)	(kJ)
9-7 E	Lot A	1.77	0.895	<5%		
11-8 E	L5(2)	1.67	0.915	60%	340	514
12-9 E	Lot A	1.84	0.877	<5%		
15-7 E	L5(1)	1.77	0.895	<5%		
16-8 E	L5(4)	1.86	0.871	<5%		
17-9 E	L1(1)	1.55	0.939	20%	460	407
21-7 E	# 830	1.92	0.836	<5%		
22-8 E	L5(3)	1.8	0.888	30%	200	188
35-7 <mark>S</mark>	L17(1)	1.32	0.985	12%	1800	553
35-8 <mark>S</mark>	L17(2)	0.95	1.059	13%	2066	313
35-9 <mark>S</mark>	L17	1.39	0.971	1%		
43-7 <mark>S</mark>	L14-2	1.73	0.903	80%	1250	245
43-8 <mark>S</mark>	ETI	1.63	0.923	5%	525	65
43-9 <mark>S</mark>	L14-3	1.61	0.927	1%		
51-7 <mark>S</mark>	L25B-1	1.55	0.939	12%	266	176
51-8 <mark>S</mark>	L25A-2	1.52	0.945	5%	133	14
51-9 <mark>S</mark>	L19	1.54	0.941	43%	79	28
56-7 <mark>S</mark>	L24F	1.55	0.939	15%	2095	536
56-8 <mark>S</mark>	L24D	1.84	0.877	4%		
56-9 <mark>S</mark>	L25B-2	1.56	0.937	3%		
57-8 <mark>S</mark>	Pd-C	N.A.	N.A.	300%	93	115
58-9 <mark>S</mark>	L25A	1.69	0.911	200%	540	485
61-7 <mark>S</mark>	L25B-1	1.63	0.923	50%	105	146
	Cell - Calorima 9-7 E 11-8 E 12-9 E 15-7 E 16-8 E 17-9 E 21-7 E 22-8 E 35-7 S 35-8 S 35-8 S 35-9 S 43-7 S 43-8 S 43-9 S 51-7 S 51-8 S 51-9 S 56-8 S 56-8 S 56-8 S 56-9 S 57-8 S 58-9 S 61-7 S	Cell - CathodeCalorimeter $9-7 E$ Lot A $11-8 E$ L5(2) $12-9 E$ Lot A $15-7 E$ L5(1) $16-8 E$ L5(4) $17-9 E$ L1(1) $21-7 E$ # 830 $22-8 E$ L5(3) $35-7 S$ L17(1) $35-8 S$ L17(2) $35-9 S$ L17 $43-7 S$ L14-2 $43-8 S$ ETI $43-9 S$ L14-3 $51-7 S$ L25B-1 $51-8 S$ L25A-2 $51-9 S$ L19 $56-7 S$ L24F $56-8 S$ L24D $56-9 S$ L25B-2 $57-8 S$ Pd-C $58-9 S$ L25A $61-7 S$ L25B-1	Cell -CathodeMin.Calorimeter R/R° 9-7 ELot A 1.77 11-8 EL5(2) 1.67 12-9 ELot A 1.84 15-7 EL5(1) 1.77 16-8 EL5(4) 1.86 17-9 EL1(1) 1.55 21-7 E# 830 1.92 22-8 EL5(3) 1.8 35-7 SL17(1) 1.32 35-8 SL17(2) 0.95 35-9 SL17 1.39 43-7 SL14-2 1.73 43-8 SETI 1.63 43-9 SL14-3 1.61 51-7 SL25B-1 1.55 51-8 SL25A-2 1.52 51-9 SL19 1.54 56-7 SL24F 1.55 56-8 SL24D 1.84 56-9 SL25B-2 1.60 57-8 SPd-CN.A.58-9 SL25B-1 1.63 61-7 SL25B-1 1.63	Cell -CathodeMin.Max.Calorimeter R/R° D/Pd 9-7 ELot A1.770.89511-8 EL5(2)1.67 0.915 12-9 ELot A1.840.87715-7 EL5(1)1.770.89516-8 EL5(4)1.860.87117-9 EL1(1)1.55 0.939 21-7 E# 8301.920.83622-8 EL5(3)1.8 0.888 35-7 SL17(1)1.32 0.985 35-8 SL17(2)0.95 1.059 35-9 SL171.390.97143-7 SL14-21.73 0.903 43-8 SETI1.630.92343-9 SL14-31.610.92751-7 SL25B-11.55 0.939 51-8 SL25A-21.52 0.945 51-9 SL191.54 0.941 56-7 SL24F1.55 0.939 56-8 SL24D1.840.87756-9 SL25B-21.560.93757-8 SPd-CN.A.N.A.58-9 SL25A-11.63 0.923	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	Cell -CathodeMin.Max.Excess PowerCalorimeter R/R° D/Pd % of P_{In} (mW)9-7 ELot A1.770.895<5%

E = Energetics and S = SRI Data Acquisition.

Energetics D/A - null heat result

-Pin-Pe-(W) - Pout(W) + Pxs(W) - Current(A) - R/R0



Energetics D/A - positive excess heat

--Pin-Pe-(W) --Pout(W) + Pxs(W) --Current(A) ---R/R0



SRI D/A - null heat result (1)



SRI D/A - null heat result (2)



SRI D/A - associated poor loading and dynamics





SRI D/A - associated high loading and dynamics



SRI D/A mode B excess heat (1)



Electrolysis duration (hours)

SRI D/A mode B excess heat (2)



Electrolysis duration (hours)



ENEA Mass Flow Calorimeter



ENEA - Results

Data	Cathode	Min.	Max.	Excess	Power
Acquisition		R/R°	D/Pd	% of P_{In}	Mode
ENEA	L14	1.54	0.941	80	В
ENEA	L17	1.4	0.969	500	В
ENEA	L19	1.7	0.909	100	А
ENEA	L23	1.69	0.911	37	В
ENEA	L25A	1.8	0.888	24	В
ENEA	L30	1.78	0.892	7000	В

Discussion (1)

Excess power was observed to have two phenomenologically different forms that we tentatively identify as Modes A and B. The general features of these different modes are as follows.
<u>Mode A</u>. This behavior conforms closely to

 $\mathbf{P}_{xs} = \mathbf{M} (\mathbf{x} - \mathbf{x}^\circ)^2 (\mathbf{i} - \mathbf{i}^\circ) |\mathbf{i}_{\mathrm{D}}|$

Where: M is a proportionality constant,

x = D/Pd is the deuterium atom loading and

x° the threshold loading below which no excess heat is observed

- typically $x^o \approx 0.875$

- i is the electrochemical current or current density and
- i° a critical threshold, and
- $|i_D|$ is the flux of D across the interface expressed as a current density.

Discussion (2)

- The interfacial flux can be calculated directly from the minimum and maximum loading values obtained from the resistance ratio measurements in the fundamental superwave interval (15 or 20 minutes).
- Probably as a result of the superwave dynamics the interfacial flux term, $|i_D|$, was up to an order of magnitude larger in the present study compared with previous dc electrolysis of palladium wires.
- This combination of factors led to excess power effects of 5-50% of the input power, in Mode A, very consistent with previous excess heat results.
- Although the proportionality constant M is not well specified it probably reflects some properties of surface heterogeneity.
- This equation permits explanation of experiments that do not produce heat excess. The failure to meet and <u>maintain</u> the current, loading, and flux criteria <u>simultaneously</u> results in a failure to observe Mode A excess heat.

Discussion

- <u>Mode B</u>. A second mode of behavior was seen in three experiments at SRI and five at ENEA, and in all of those exhibiting excess power greater than 100%.
- This mode is more typical of that reported previously by Energetics.
 - Mode B excess heat initiates within 6 h of the application of cathodic current (or 4 h of maximum loading), whereas Mode A behavior requires a longer initiation time, typically several hundreds of hours.
 - ii) Mode B excess heat responds sluggishly to input cathodic current density and, so far, exhibits no obvious current density threshold.
 - iii) Mode B excess heat has not been observed at D/Pd loadings less than the threshold typical of Mode A behavior (D/Pd ≈ 0.875) but appears to respond only transiently to increased average loading.

Conclusions

- The Energetics calorimeters and cells were found to be well designed and calibrated, and capable of steady baseline operation in the absence of excess heat.
- The three sigma (3σ) calorimetric uncertainty was estimated to be approximately 5% of the input power under normal input conditions.
- Of the fifteen experiments performed using SRI data acquisition, eleven produced excess heat at or above the 3 σ experimental uncertainty.
- This high level of reproducibility is attributable to two conspicuous differences between Energetics experiments and all those that preceded them:
 - i. Very high deuterium atom loadings that result from SuperWave[®] cathodization of appropriately prepared palladium foils.
 - ii. The extraordinarily high interfacial flux of deuterium in and out through the palladium cathode surface that results from superwave stimulus.
 - iii. Although more reproducible the results of the Energetics/ENEA collaboration are consistent with those that preceded them:

Excess Power Production vs. Maximum loading of cathode

